## **Palladium-Catalyzed Arylation of Enynes and Electron-Deficient Alkynes Using Diaryliodonium Salts**

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**ABSTRACT**

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**A new single-pot procedure for the synthesis of aryl alkynes is described. Palladium catalyzes the coupling reaction of diaryliodonium compounds with enynes and electron-deficient alkynes to give aryl alkynes in good yields.**

The palladium-catalyzed coupling of terminal alkynes with aryl iodides to give arylalkynes is an important reaction in organic chemistry (Sonogashira coupling).<sup>1,2</sup> Alkynes obtained in these reactions are useful building blocks in organic and materials chemistry.<sup>3</sup> However, this reaction has limitations. Alkynes containing an electron-withdrawing group (directly attached to the ethynyl carbon) do not react significantly with aryl iodides;<sup>4</sup> therefore, not many reports appear in the literature.<sup>5</sup> Also, palladium-catalyzed arylation of enynes occurs only in moderate yields.<sup>6</sup> Herein, we report a new, efficient, and mild palladium-catalyzed method for the arylation of enynes and electron-deficient alkynes using diaryliodonium salts.

(5) Eckert, T.; Ipaktschi, J. *Synth. Commun.* **<sup>1998</sup>**, *<sup>28</sup>*, 327-335.

Iodonium compounds are widely available and generally prepared under mild conditions.7 Aryliodonium salts show superior reactivity compared to aryl halides. The highly electron withdrawing nature of the  $PhI^+$  moiety in iodonium compounds activates the carbon-iodine bond toward various reactions.7 Diaryliodonium salts **1a** and **1b** were prepared by a literature procedure.<sup>8</sup>

First, we investigated the coupling of a diaryliodonium compound (**1a**) with a methyl enyne (**2a**) in the presence of palladium as a catalyst and copper as a cocatalyst. Different palladium sources and solvents were examined for this coupling reaction; the  $PdCl_2(PPh_3)_2/CuI/K_2CO_3$  system was found to be superior to other combinations. With this catalytic system, reaction occurred at rt to give essentially a quantitative yield of the aryl-substituted enyne **3a** using either triflate or tosylate as the counterion (Table 1, entries 1 and 2).

Terminal alkynes containing an electron-withdrawing group directly attached to the ethynyl carbon atom do not react significantly with aryl halides. Therefore, generally an alternative methodology is used to prepare aryl propiolic

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<sup>(3) (</sup>a) *Modern Acetylene Chemistry*; Stang, P. J., Diederich, F., Eds.; VCH: Weinheim, 1995. (b) Brandsma, L.; Vasilevsky, S. F.; Verkruijsse, H. D. *Application of Transition Metal Catalysts in Organic Synthesis*; Springer: Berlin, 1998; p 198.

<sup>(4) (</sup>a) Yoneda, N.; Matsuoka, S.; Miyaura, N.; Fukuhara, T.; Suzuki, A. *Bull. Chem. Soc. Jpn.* **1990**, 63, 2124–2126. (b) Sakamoto, T.; Shiga, A. *Bull. Chem. Soc. Jpn.* **<sup>1990</sup>**, *<sup>63</sup>*, 2124-2126. (b) Sakamoto, T.; Shiga, F.; Yasuhara, A.; Uchiyama, D.; Kondo, Y.; Yamanaka, H. *Synthesis* **1992**, <sup>746</sup>-748. (c) Kundu, N. G.; Dasgupta, S. K. *J. Chem. Soc., Perkin Trans. <sup>1</sup>* **<sup>1993</sup>**, 2657-2663.

<sup>(6)</sup> Iwasawa, N.; Satoh, H. *J. Am. Chem. Soc.* **<sup>1999</sup>**, *<sup>121</sup>*, 7951-7952.

<sup>(7) (</sup>a) Stang, P. J.; Zhdankin, V. V. *Chem. Re*V*.* **<sup>1996</sup>**, *<sup>96</sup>*, 1123-1178. (b) Varvoglis, A. *Hypervalent Iodine in Organic Synthesis*; Academic<br>Press: London 1997 (c) Wirth T: Hirt U H. *Synthesis* 1999 1271– Press: London, 1997. (c) Wirth, T.; Hirt, U. H. *Synthesis* **<sup>1999</sup>**, 1271- 1287.

<sup>(8) (</sup>a) Koser, G. F.; Wettach, R. H.; Smith, C. S. *J. Org. Chem.* **1980**, *<sup>45</sup>*, 1543-1544. (b) Stang, P. J.; Zhdankin, V. V.; Tykwinski, R.; Zefirov, N. S. *Tetrahedron Lett.* **<sup>1991</sup>**, *<sup>32</sup>*, 7497-7498.



esters<sup>4b</sup> and acetylenic ketones.<sup>4c,10</sup> In contrast, the aryliodonium salts underwent coupling with alkynes containing electron-withdrawing groups such as ester **2b** and ketones **2c**, <sup>11</sup> **2d**, and **2e** to give arylated products in good yield (Table 1, entries  $3-6$ ).

It was also of interest to study this reaction using unsymmetrically substituted iodonium salts. Aryliodonium **1c** was obtained using a modified literature procedure.12

(9) **Typical procedure:** To a mixture of  $PdCl_2(PPh_3)_2$  (0.021 g, 0.03) mmol),  $\mathbf{1a}$  (0.643 g, 1.5 mmol), CuI (0.04 mmol), and  $K_2CO_3$  (0.276 g, 2 mmol) in DMF/H2O (6:1) (14 mL) was added enyne **2a** (0.099 g, 1.5 mmol) at rt under an  $N_2$  atmosphere, and the solution was stirred at rt for  $2-3$  h. The reaction was quenched by an aqueous NH4Cl solution; workup and column chromatography (silica gel, hexane) gave **3a** (0.21 g, 99%). (10) Tohda, Y.; Sonagashira, K.; Hagihara, N. *Synthesis* **<sup>1977</sup>**, 777-

778. (11) Muzart, J.; Ajjou, A. N. A. *Synthesis* **<sup>1993</sup>**, 785-787.

(12) Kuehl, C. J.; Bolz, J. T.; Zhdankin, V. V. *Synthesis* **<sup>1995</sup>**, 312- 316.

Coupling of iodonium salt **1c** and methyl enyne **2a** proceeded smoothly to give product **3f** (70%) (Scheme 1). The phenyl



ring with the electron-donating group (TMS) underwent coupling to give the corresponding arylated product.

A plausible mechanism for this coupling reaction is depicted in Scheme 2. The reaction most likely begins with



the oxidative addition of  $Pd<sup>0</sup>$  to the iodonium compound to give arylpalladium species **4** which on reaction with copper acetylide leads to the aryl(alkynyl)palladium species **5** that on extrusion of Pd0 leads to the formation of aryl alkyne **3**. Both the palladium and copper(I) iodide were essential catalysts for this reaction. The aryl iodides obtained as byproduct lends support to our mechanism.

In summary, we have developed a new, superior, and mild protocol for the arylation of enynes and electron-deficient alkynes using aryliodonium salts as the electrophilic coupling partners, where aryl iodides either do not work or give only low yields of products.

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**Supporting Information Available:** A description of experimental details and <sup>1</sup>H NMR spectra of the compounds **3d**, **3e**, and **3f**. This material is available free of charge via the Internet at http://pubs.acs.org.

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