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Palladium-Catalyzed Arylation of Enynes and Electron-Deficient Alkynes Using Diaryliodonium Salts

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ABSTRACT

A new single-pot procedure for the synthesis of aryl alkynes is described. Palladium catalyzes the coupling reaction of diaryliodonium compounds with enynes and electron-deficient alkynes to give aryl alkynes in good yields.

The palladium-catalyzed coupling of terminal alkynes with aryl iodides to give arylalkynes is an important reaction in organic chemistry (Sonogashira coupling).^{1,2} Alkynes obtained in these reactions are useful building blocks in organic and materials chemistry.³ However, this reaction has limitations. Alkynes containing an electron-withdrawing group (directly attached to the ethynyl carbon) do not react significantly with aryl iodides;⁴ therefore, not many reports appear in the literature.⁵ Also, palladium-catalyzed arylation of enynes occurs only in moderate yields.⁶ Herein, we report a new, efficient, and mild palladium-catalyzed method for the arylation of enynes and electron-deficient alkynes using diaryliodonium salts.

Iodonium compounds are widely available and generally prepared under mild conditions.⁷ Aryliodonium salts show superior reactivity compared to aryl halides. The highly electron withdrawing nature of the PhI⁺ moiety in iodonium compounds activates the carbon—iodine bond toward various reactions.⁷ Diaryliodonium salts **1a** and **1b** were prepared by a literature procedure.⁸

First, we investigated the coupling of a diaryliodonium compound (1a) with a methyl enyne (2a) in the presence of palladium as a catalyst and copper as a cocatalyst. Different palladium sources and solvents were examined for this coupling reaction; the PdCl₂(PPh₃)₂/CuI/K₂CO₃ system was found to be superior to other combinations. With this catalytic system, reaction occurred at rt to give essentially a quantitative yield of the aryl-substituted enyne 3a using either triflate or tosylate as the counterion (Table 1, entries 1 and 2).

Terminal alkynes containing an electron-withdrawing group directly attached to the ethynyl carbon atom do not react significantly with aryl halides. Therefore, generally an alternative methodology is used to prepare aryl propiolic

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Table 1. Palladium-Catalyzed Arylation of Alkynes⁹

entry	Х	Υ	Product	Yield ^a
1	OTf 1a	(CH ₃)C=CH ₂ 2a	<u> </u>	99%
2	OTs 1b	(CH ₃)C=CH ₂ 2a	<u> </u>	99%
3	OTf 1a	-COOEt	COOEt 3b	80%
4	OTs 1b	-CO-Ph 2c	\bigcirc	70%
5	OTs 1b	-CO-cyhex 2d	Cyhex	65%
6	OTf 1a	-CO-Ad 2e	Ad Ad	71%

^a Isolated yields: cyhex, cyclohexane, Ad, adamantane.

esters^{4b} and acetylenic ketones.^{4c,10} In contrast, the aryliodonium salts underwent coupling with alkynes containing electron-withdrawing groups such as ester **2b** and ketones **2c**, ¹¹ **2d**, and **2e** to give arylated products in good yield (Table 1, entries 3–6).

It was also of interest to study this reaction using unsymmetrically substituted iodonium salts. Aryliodonium 1c was obtained using a modified literature procedure.¹²

Coupling of iodonium salt **1c** and methyl enyne **2a** proceeded smoothly to give product **3f** (70%) (Scheme 1). The phenyl

Scheme 1. Unsymmetrical Iodonium-Enyne Coupling

F₅

1c
SiMe₃

$$\frac{Pd(PPh_3)_2Cl_2, Cul}{K_2CO_3, DMF/H_2O}$$
Me₃Si
$$\frac{3f}{70\%}$$

ring with the electron-donating group (TMS) underwent coupling to give the corresponding arylated product.

A plausible mechanism for this coupling reaction is depicted in Scheme 2. The reaction most likely begins with

Scheme 2. Proposed Mechanism of Coupling Reactions

the oxidative addition of Pd^0 to the iodonium compound to give arylpalladium species 4 which on reaction with copper acetylide leads to the aryl(alkynyl)palladium species 5 that on extrusion of Pd^0 leads to the formation of aryl alkyne 3. Both the palladium and copper(I) iodide were essential catalysts for this reaction. The aryl iodides obtained as byproduct lends support to our mechanism.

In summary, we have developed a new, superior, and mild protocol for the arylation of enynes and electron-deficient alkynes using aryliodonium salts as the electrophilic coupling partners, where aryl iodides either do not work or give only low yields of products.

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Supporting Information Available: A description of experimental details and ¹H NMR spectra of the compounds **3d**, **3e**, and **3f**. This material is available free of charge via the Internet at http://pubs.acs.org.

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⁽⁹⁾ **Typical procedure:** To a mixture of $PdCl_2(PPh_3)_2$ (0.021 g, 0.03 mmol), **1a** (0.643 g, 1.5 mmol), CuI (0.04 mmol), and K_2CO_3 (0.276 g, 2 mmol) in DMF/H₂O (6:1) (14 mL) was added enyne **2a** (0.099 g, 1.5 mmol) at rt under an N_2 atmosphere, and the solution was stirred at rt for 2–3 h. The reaction was quenched by an aqueous NH_4Cl solution; workup and column chromatography (silica gel, hexane) gave **3a** (0.21 g, 99%).

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